

# Epsonwf7511adjustmentprogramdownloadhit 5 ((NEW))

Wii ChatWindowingfilter.ini Out.to wiki was the manual with the same name. (a) (3) . . . . (b) pdf files with these extensions or names: . . . . . The present invention relates to electrochemical cells and, more particularly, to an electrochemical cell that provides a low internal resistance and a high electrical capacity in the range of 100 to 150 milliampere-hour at a high operating temperature. The secondary electrolyte for alkaline cells, particularly those containing zinc, is commonly an inorganic oxide, such as zinc oxide, zinc hydroxide, zinc carbonate, and the like. The additives which normally are used to prepare the inorganic oxide include organic compounds, such as ethers and esters. In developing the art of alkaline secondary batteries, it is desirable to employ a secondary active material, such as zinc, which is relatively inexpensive and easy to handle on a commercial scale. One means by which a large quantity of zinc material can be economically produced involves the electrolytic reduction of metal in a molten alkaline electrolyte. Numerous efforts have been made to provide an electrolyte for a zinc secondary cell that would be low in cost and provide secondary material in a form that is easy to use. It is important that the secondary material be dry and stable during storage and use. The electrolyte should be one that forms an excellent and durable seal with the container or housing of the cell. One such electrolyte is disclosed in U.S. Pat. No. 2,718,126. This patent discloses a molten sodium potassium zirconate electrolyte formed in situ from a zirconium salt and a potassium salt. The electrolyte is prepared by mixing a 1:1 zirconium potassium hydroxide solution with an aqueous solution of potassium hydroxide. The zirconium potassium hydroxide solution is prepared by adding potassium hydroxide to an aqueous solution of zirconyl chloride to form the zirconium salt. U.S. Pat. No. 3,282,875 discloses electrolyte mixtures for primary or secondary alkaline batteries. An alkali metal or ammonium carbonate in an amount of at least about 4% of the total alkali metal or ammonium carbonate is mixed with an alkali



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